

# Design of activated serine-containing catalytic triads with atomic-level accuracy

Sridharan Rajagopalan<sup>1,8</sup>, Chu Wang<sup>2,8</sup>, Kai Yu<sup>1</sup>, Alexandre P Kuzin<sup>3</sup>, Florian Richter<sup>1,4,7</sup>, Scott Lew<sup>3</sup>, Aleksandr E Miklos<sup>5</sup>, Megan L Matthews<sup>2</sup>, Jayaraman Seetharaman<sup>3</sup>, Min Su<sup>3</sup>, John F Hunt<sup>3</sup>, Benjamin F Cravatt<sup>2</sup> & David Baker<sup>1,4,6\*</sup>

**A challenge in the computational design of enzymes is that multiple properties, including substrate binding, transition state stabilization and product release, must be simultaneously optimized, and this has limited the absolute activity of successful designs. Here, we focus on a single critical property of many enzymes: the nucleophilicity of an active site residue that initiates catalysis. We design proteins with idealized serine-containing catalytic triads and assess their nucleophilicity directly in native biological systems using activity-based organophosphate probes. Crystal structures of the most successful designs show unprecedented agreement with computational models, including extensive hydrogen bonding networks between the catalytic triad (or quartet) residues, and mutagenesis experiments demonstrate that these networks are critical for serine activation and organophosphate reactivity. Following optimization by yeast display, the designs react with organophosphate probes at rates comparable to natural serine hydrolases. Co-crystal structures with diisopropyl fluorophosphate bound to the serine nucleophile suggest that the designs could provide the basis for a new class of organophosphate capture agents.**

Serine hydrolases use a conserved nucleophilic serine to hydrolyze ester, amide or thio-ester bonds in proteins and small molecules and constitute one of the largest enzyme families in the human proteome<sup>1</sup>. The canonical serine-histidine-aspartic acid catalytic triad mechanism has been well studied: the serine Oγ is the nucleophile, the imidazole ring of the histidine acts as a general acid/base, and the carboxylate of the aspartic acid orients the imidazole ring and neutralizes the charge developed on the histidine in the transition state. In addition to the catalytic triad residues, a highly conserved oxyanion-binding site, commonly referred to as the oxyanion hole, stabilizes the negative charge on the carbonyl oxygen of the tetrahedral intermediate. The peptide backbone NH groups stabilize these high-energy species in most cases<sup>2-7</sup>. The activated nucleophilic serine can be covalently modified by electrophilic organophosphate compounds, resulting in the loss of catalytic activity. This has inspired the development of fluorophosphonates as activity-based probes for serine hydrolases, which enable monitoring of the activity of many serine hydrolases in parallel directly in complex proteomes by the activity-based protein profiling (ABPP) technology<sup>8,9</sup>.

To make progress in enzyme design, it is useful to focus on critical aspects of catalysis (in this case, the active site nucleophile that initiates catalysis) independent of full multiple turnover reaction cycles, whose optimization requires many complex tradeoffs; such a reduction in complexity can facilitate the solution of very challenging problems. Recently, we designed esterases with a cysteine-histidine dyad and an oxyanion hole in the active site<sup>10</sup>. Though these designs contained active site cysteines with heightened nucleophilicity<sup>11</sup> and were catalytically active, crystal structures revealed that the active site histidine residues, which were intended to activate the cysteine nucleophiles, were not properly positioned<sup>10</sup>. Catalytic activity was most likely achieved because cysteine, with a pK<sub>a</sub> around 8, is one

of the most intrinsically nucleophilic amino acids, and its inherent reactivity can be enhanced without specific interactions with supporting residues in the enzyme's active site. Compared to cysteine and lysine, which was used in *de novo* designed retroaldolases<sup>12</sup>, serine has a much higher pK<sub>a</sub> at ~13, and hence its nucleophilicity typically depends on interactions with activating residues such as the histidine and the aspartate or glutamate of the catalytic triad. It is therefore methodologically more challenging to design serines with heightened nucleophilicity: success most likely depends on accurate designed hydrogen bonding interactions with an activating residue (or residues).

Here we take on the challenge of designing active sites with serines with nucleophilicity approaching that found in the catalytic triads of native hydrolases. *De novo* designed proteins with activated serine nucleophiles could be useful in their own right as scavengers of organophosphate nerve agents. We use computational methods to design proteins with idealized serine-containing catalytic triads and take advantage of activity-based fluorophosphonate probes<sup>13</sup> to screen and assess the serine nucleophilicity of these designs directly in native cellular systems.

## RESULTS

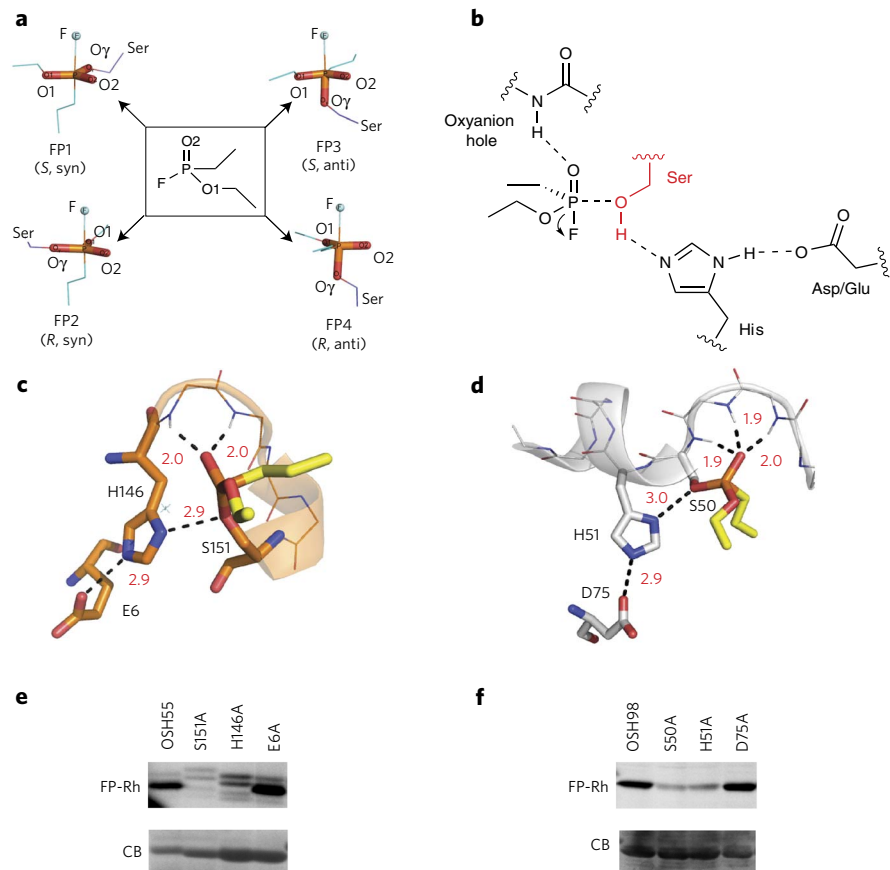
### Computational design of catalytic triad active sites

We selected a fluorophosphonate-reactive group (Fig. 1a) as the model organophosphate substrate because it provides (i) a suitably tempered electrophile to test the serine nucleophilicity of protein designs and (ii) a mimic of the tetrahedral transition state that occurs during serine hydrolase catalysis. From a technical perspective, reporter-tagged fluorophosphonates should provide a straightforward way to screen enzyme designs in complex proteomes, as we have shown for other ABPP probes<sup>11</sup>. Four transition state models corresponding to *syn* and *anti* attack for two fluorophosphonate

<sup>1</sup>Department of Biochemistry, University of Washington, Seattle, Washington, United States. <sup>2</sup>The Skaggs Institute for Chemical Biology and Department of Chemical Physiology, The Scripps Research Institute, La Jolla, California, United States. <sup>3</sup>Northeast Structural Genomics Consortium, Department of Biological Sciences, Columbia University, New York, New York, USA. <sup>4</sup>Graduate Program in Biological Physics, Structure and Design, University of Washington, Seattle, Washington, USA. <sup>5</sup>EXCET, Inc., Springfield, Virginia, USA. <sup>6</sup>Howard Hughes Medical Institute, University of Washington, Seattle, Washington, USA. <sup>7</sup>Present address: Institute of Biology, Humboldt-Universität zu Berlin, Germany. <sup>8</sup>These authors contributed equally to this work.

\*e-mail: dabaker@uw.edu

**Figure 1 | Design strategy.** (a) Transition state models. The fluorophosphonate (FP) ligand (ethyl ethylphosphonofluoridate) used in design calculations is in the middle panel flanked by transition state models corresponding to *syn* and *anti* attack by the serine (purple) nucleophile O $\gamma$  on the two isomers (*R* and *S*) of the ligand with the leaving atom fluorine (F) in blue spheres. (b) Theozyme geometry. Geometric parameters were derived from the computed ideal active site geometries and native hydrolase statistics previously reported<sup>30</sup>. Theozymes with the histidine hydrogen bonding reversed were also considered during RosettaMatch (Online Methods). (c,d) Design models of OSH55 (c) and OSH98 (d). Active site residues of OSH55 and OSH98 are shown in orange and white sticks, respectively. The modeled fluorophosphonate ligand is shown in yellow sticks. In both designs, multiple backbone NHs form the oxyanion hole to stabilize the transition state. (e,f) Selective FP-Rh labeling of OSH55 (e) and OSH98 (f) but not the active site serine knockouts. Serine reactivity was assessed by gel-based ABPP (top, in-gel fluorescence; bottom, Coomassie blue (CB) staining; full-size gel images are in **Supplementary Fig. 15**). The experiments were performed in duplicate with consistent results.



isomers (*R* and *S*) were generated using bond angles and bond lengths obtained from quantum mechanics/molecular mechanics (QM/MM) calculations (**Fig. 1a** and **Supplementary Results, Supplementary Table 1**)<sup>14</sup>. Ideal active sites with the transition state model, a serine-histidine-aspartate/glutamate catalytic triad, and a backbone oxyanion hole in orientations optimal for catalysis (**Fig. 1b**) were built as described in the Online Methods. We used RosettaMatch to search for the placement of these 'theozymes' in a set of 800 protein scaffolds<sup>15</sup>, none of which contained an existing catalytic triad-based active site. The residues surrounding the matched sites were then optimized by RosettaDesign<sup>16</sup> to stabilize the active site residues, increase affinity for the fluorophosphonate ligand, or both. Eighty-five designs were chosen for experimental characterization on the basis of the accuracy of the active site geometry following unconstrained combinatorial side chain optimization and the transition state binding energy (Online Methods). The overall design workflow is illustrated in **Supplementary Figure 1a**, and the scaffolds on which the 85 designs are based are listed in **Supplementary Table 2**.

### Experimental screening of designs

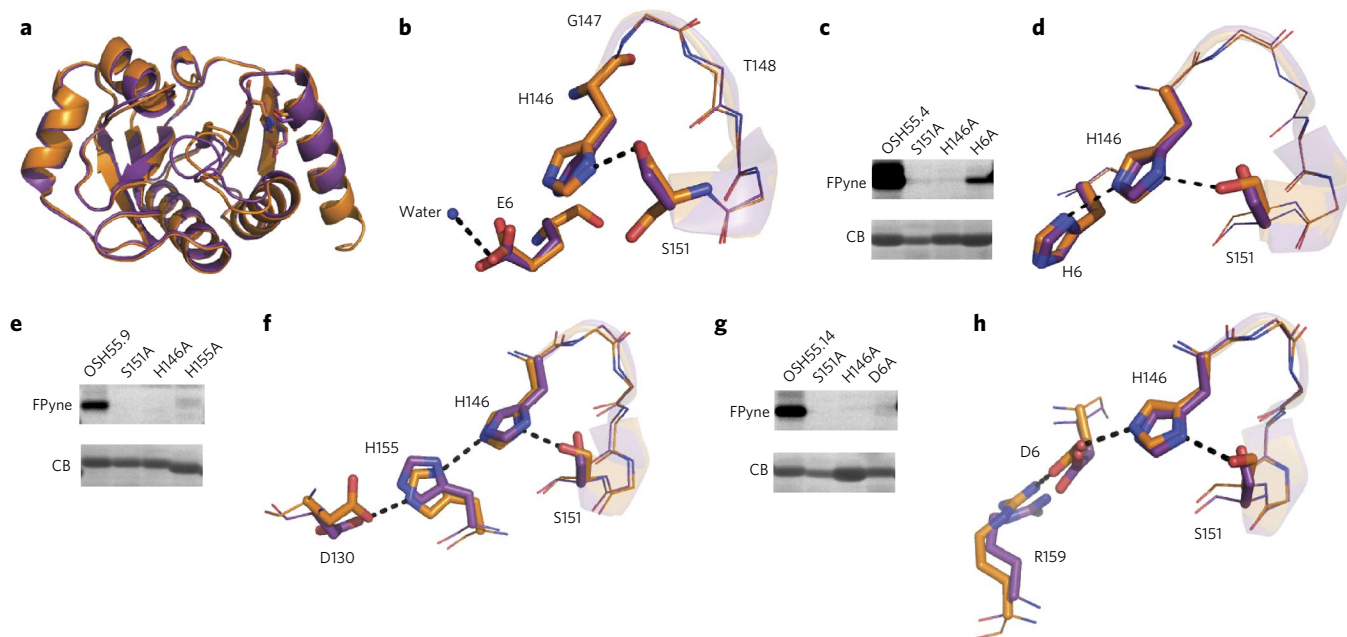
Genes corresponding to the 85 designs were ordered and expressed in *Escherichia coli*. Roughly half of the designs did not express as soluble proteins; the requirement that the active site be preformed favored designs with the catalytic triad aspartic acid or glutamic acid at least partially buried, and this may have destabilized many of the proteins. The reactivity of the active site serine in the soluble designs was assessed in cell lysates by gel-based ABPP (**Supplementary Fig. 1b**) using fluorophosphonate-rhodamine (FP-Rh) and fluorophosphonate-alkyne (FPyne) probes (**Supplementary Fig. 2**). Two of the designs, OSH55 and OSH98 (**Fig. 1c,d**), showed modest reactivity with the fluorophosphonate probes. The catalytic triads in the designs involve all non-native residues; for example, in OSH55, the catalytic triad residues Ser151, His146 and Glu6 are respectively aspartic acid, glycine and valine in the native scaffold (Protein Data Bank

(PDB) code 2DX6; **Supplementary Fig. 3**). Mutagenesis of the designed active site serine and histidine residues to alanine indicated that probe labeling was specific to the designed serines (**Fig. 1e,f**). OSH55 was chosen for further characterization because of its small size (163 amino acids) and the stability of the starting scaffold. In the design model, OSH55 has two backbone amide protons contributing to the oxyanion hole (**Fig. 1c**).

### Characterization and optimization of OSH55

We solved the structure of OSH55 by crystallography at 2.3-Å resolution (PDB code 3V45). The design model and crystal structure match closely with an overall backbone r.m.s. deviation of 0.6 Å (**Fig. 2a**). The crystal structure conformations of the active site Ser151 and His146 side chains and the backbone geometry of the oxyanion hole, composed of the backbone NH of residues Gly147 and Thr148, are nearly identical to those in the design model (**Fig. 2b**). Glu6, designed to stabilize His146, moves slightly away from the position in the design model and interacts with a nearby water molecule (**Fig. 2b**). Consistent with this observation, mutation of Glu6 did not affect fluorophosphonate probe labeling (**Fig. 1e**), suggesting that it does not contribute to activation of the active site serine.

Because the crystal structure and mutagenesis experiments indicated that Glu6 was not interacting with His146 as it did in the model, we used computational design to identify alternative solutions with histidine, aspartic acid or glutamic acid residues positioned to hydrogen bond to and stabilize the desired rotamer conformation of His146 (**Supplementary Table 3**). The alternative catalytic triad designs were expressed and tested for fluorophosphonate reactivity by ABPP (**Supplementary Fig. 4**). Three designs were found to have markedly increased reactivity toward the FPyne probe. In OSH55.4, the third catalytic triad residue is His6. Knockout of any of the three catalytic triad residues abolished FPyne labeling, suggesting that, in the Ser151-His146-His6 triad, the



**Figure 2 | Design and experimental characterization of OSH55 and related designs.** (a) Structural superposition of design (brown) and crystal structure (purple) of OSH55 with overall backbone r.m.s. deviation of 0.6 Å. The all-atom r.m.s. deviation over the binding pocket residues is 0.34 Å. (b) A zoom-in view of the active site of OSH55 showing the orientation of Ser151 and His146 as designed and Glu6 coordinating a water molecule. (c,e,g) Selective fluorophosphonate probe labeling of OSH55.4 (c), OSH55.9 (e) and OSH55.14 (g) but not the active site knockout mutants. Serine reactivity was assessed by gel-based ABPP (top, in-gel fluorescence; bottom, Coomassie blue (CB) staining; full-size gel images are in **Supplementary Fig. 16**). The experiments were performed in duplicate with consistent results. There is much less labeling in the catalytic triad knockouts but no decrease in the amount of protein. (d,f,h) Superimposition of the designed catalytic triad active sites (brown) overlaid on the experimentally determined crystal structures (purple) for OSH55.4 (d), OSH55.9 (f) and OSH55.14 (h); the designed hydrogen bond networks are recapitulated with very high accuracy.

nucleophilicity of Ser151 is substantially activated by the supporting His-His dyad (Fig. 2c,d). LC/MS/MS confirmed that the FPpyne is covalently bound to Ser151 in OSH55.4 (**Supplementary Fig. 5**). OSH55.9 also contains a serine-histidine-histidine triad in the designed active site, but the third histidine comes from a different position (A155H); mutagenesis of the serine again abolished FPpyne labeling (Fig. 2e,f). Serine-histidine-histidine triads are less common than serine-histidine-aspartic acid triads but are found in some native serine hydrolases, such as herpes virus peptidases<sup>3,17</sup>. OSH55.14 has a more canonical catalytic triad, generated by replacing His6 with an aspartic acid, and it incorporates an additional arginine (L159R) to stabilize the Asp6 side chain orientation in the catalytic triad. Mutagenesis of the catalytic triad histidine and aspartic acid in addition to the serine nucleophile abolished FPpyne labeling (Fig. 2g,h). Thus, as in OSH55.4, both the histidine and the most distal residue in the triad (histidine in OSH55.4 and OSH55.9, aspartic acid in OSH55.14) contribute to serine activation, suggesting that the catalytic triad functioned as designed.

We solved the structures of OSH55.4 (PDB code 4ETJ), OSH55.9 (PDB code 4ETK) and OSH55.14 (PDB code 4ESS) to determine whether the triads had the designed conformations. In all three cases, there was close agreement between the crystal structures and design models (Fig. 2d,f,h). Most notably, in the crystal structure of OSH55.14, the designed hydrogen-bonding network Ser151-His146-Asp6-Arg159 is almost identical to that in the design model (Fig. 2h). This is perhaps the most extensive designed hydrogen bonding network that has been structurally confirmed to date and certainly the most accurate network in a functional protein design.

### Insights from structures of inactive designs

To gain insight into the factors that contribute to serine activation, we solved the structures of three designs that expressed at high levels but showed little probe labeling (**Supplementary Fig. 6**).

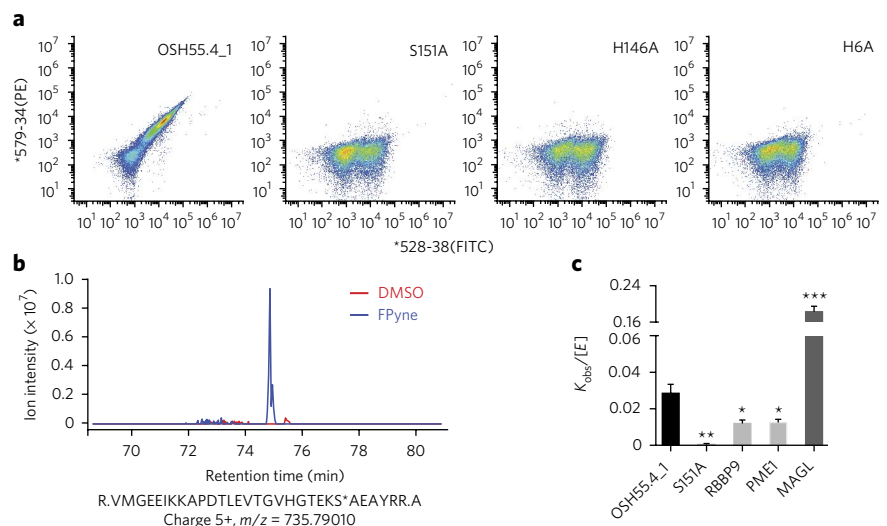
In the crystal structure of OSH97, the conformations of the designed active site residues (Ser70, His37 and Glu320) are very close to those in the design model. However, the oxyanion hole is disrupted by conversion of the loop conformation of Gly113 in the design model into a  $\beta$ -strand in the crystal structure. The lack of activity with a perfect catalytic triad but no oxyanion hole suggests that the oxyanion hole is important for transition state stabilization in the reaction with organophosphates. In the other two inactive designs (OSH26 and OSH49), there are significant backbone conformation changes in loop regions containing the catalytic triad residues which disrupt the designed sites. Comparison with the active OSH55 series of designs suggests that regular secondary structure elements may be better starting points for building up a new functional site than loop regions, as the energy gap between the designed and undesired alternative conformations will generally be larger.

To further investigate the importance of active site preorganization, we computed the average Boltzmann weight of the designed configurations of the catalytic triad residues in each of the 85 original designs and several native enzymes (**Supplementary Fig. 7**). The predicted occupancies of the designed catalytic triad conformations were much lower (<5%) in the inactive designs than in the native enzymes and the active designs. Together with the structures of the inactive designs, this suggests that lack of active site preorganization may be the primary flaw in many of the inactive designs and that future design efforts should focus on creating preorganized active sites.

### Evolutionary optimization of OSH55.4

To determine whether the fluorophosphonate reactivity of designs with structurally intact catalytic triads could be further increased, we used yeast display selection<sup>18</sup> with a biotinylated fluorophosphonate (FP-biotin) probe (**Supplementary Fig. 2**). We began with OSH55.4 because it showed the greatest fluorophosphonate

**Figure 3 | Organophosphate reactivity of OSH55.4\_1 is comparable to that of native enzymes.** (a) Flow cytometry analysis of OSH55.4\_1 and active site knockouts displayed on yeast following incubation with FP-biotin. The signal along the x axis indicates the extent of yeast surface display; the signal along the y axis shows the extent of probe binding. The active site residue knockouts abolish binding to FP-biotin. (b) MS1 chromatographic traces of the OSH55.4\_1 active site peptide adducted with the FP<sub>pyne</sub> probe. LC/MS/MS experiments (Online Methods) identified a fully tryptic peptide (sequence delimited between the two dots;  $m/z = 735.79010$  and  $z = 5+$ ) with probe modification on the catalytic Ser151 (denoted with an asterisk). The extracted MS1 chromatographic traces ( $\pm 15$  p.p.m.) showed a distinct peak present only in the FP<sub>pyne</sub>-treated (blue) but not in the DMSO-treated (red) sample. (c) Comparison of rates of FP-Rh labeling of the OSH55.4\_1 design to those of a representative set of native serine hydrolases by fluorescence polarization. The  $k_{\text{obs}}/[E]$  (mean  $\pm$  s.e.m.) value for OSH55.4\_1 is greater than that of RBBP9 and PME1. The S151A knockout mutation reduces the reaction rate to near background level. \* $P < 0.05$ , \*\* $P < 0.01$ , \*\*\* $P < 0.001$ , as determined by Student's  $t$ -test in comparison to OSH55.4\_1.



reactivity (Supplementary Fig. 4). Libraries were created by randomizing six positions around the designed active site. Four rounds of fluorescence-assisted cell sorting enrichment for increased fluorophosphonate reactivity led to two populations of cells with distinct low and high signals (Supplementary Fig. 8). The sequences of 50 clones from the high-reactivity population revealed strong convergence to a single unique sequence, OSH55.4\_1, whereas sequences from the low-reactivity pool converged to three distinct sequences, OSH55.4\_2, OSH55.4\_4 and OSH55.4\_6. Mutagenesis of each of the three catalytic triad residues of OSH55.4\_1 eliminated fluorophosphonate reactivity on yeast, confirming that the designed active site remains functional during evolution (Fig. 3a). Of the six mutations introduced during evolution, four (A112S, G113W, L126H and A155Y) were found to contribute to increased fluorophosphonate reactivity (Supplementary Fig. 9). Unlike the parent OSH55.4, which preferentially reacts with the FP<sub>pyne</sub> probe (most likely because of clashes of the bulkier rhodamine and biotin probes with an active site-proximal loop), OSH55.4\_1 has strong reactivity with all of the tested fluorophosphonate probes (FP<sub>pyne</sub>, FP-Rh and FP-biotin; Supplementary Fig. 10). LC/MS/MS confirmed that the site of fluorophosphonate labeling is the designed catalytic residue Ser151 (Fig. 3b). Mutagenesis of the serine eliminated most, but not all, of the FP-Rh labeling (Supplementary Fig. 10; the residual labeling is mostly due to Ser112 (Supplementary Fig. 11)). Mutation of the other two catalytic triad residues (H146A or H6A) completely eliminated probe labeling (Supplementary Fig. 10).

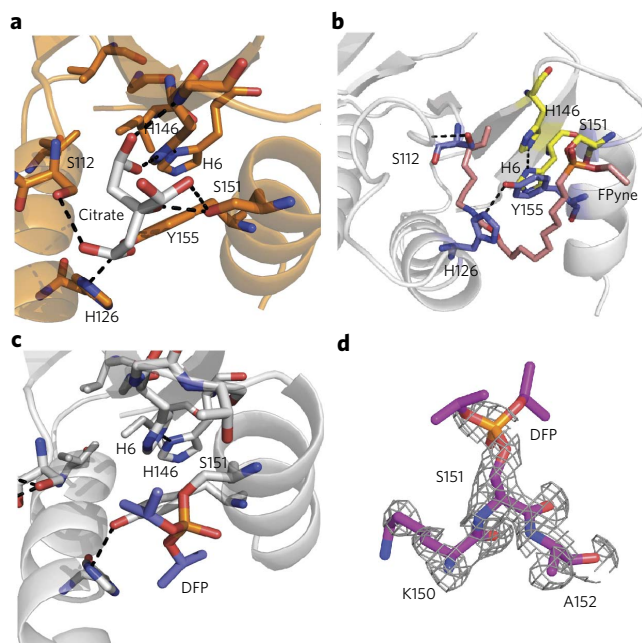
#### Relative FP reactivity of designs and natural hydrolases

We next compared the rate of fluorophosphonate labeling of optimized designs to those of a representative set of natural serine hydrolases using a fluorescence polarization (fluopol) assay<sup>19</sup>. We tested OSH55.4\_1 along with three natural serine hydrolases, retinoblastoma-binding protein 9 (RBBP9, 21.0 kDa), monoacylglycerol lipase (MAGL, 33.2 kDa) and protein phosphatase methyltransferase 1 (PME1, 42.3 kDa), all of which contain a canonical serine-histidine-aspartic acid triad active site but act on distinct classes of substrates. We monitored the changes in fluorescence polarization over time following the addition of the FP-Rh probe (0.1  $\mu\text{M}$ ) to protein samples ranging from 0.5  $\mu\text{M}$  to 5  $\mu\text{M}$  (Supplementary Fig. 12a–e). The ratios of the fitted rate constants to enzyme concentration,  $k_{\text{obs}}/[E]$  values, are useful measures of

serine reactivity. MAGL had the fastest labeling kinetics, perhaps because the probe resembles its natural fatty acylated substrate and hence fits well into the substrate binding site<sup>20</sup>. The next most reactive protein was OSH55.4\_1, which is much smaller (17 kDa) and has a more open active site. It had a notably faster fluorophosphonate labeling rate than PME1 and RBBP9 (Fig. 3c). As expected, the S151A mutant of OSH55.4\_1 showed negligible reactivity with FP-Rh in the fluopol assay (Supplementary Fig. 12b). These data suggest that, after evolutionary optimization by yeast display, the fluorophosphonate reactivity of OSH55.4\_1 is comparable to those of natural serine hydrolases.

#### Structural characterization of OSH55.4\_1

To investigate how the mutations introduced by yeast display improve fluorophosphonate reactivity, we determined the structure of OSH55.4\_1 in both the apo (PDB code 4JCA) and FP<sub>pyne</sub>-bound states (PDB code 4JLL). The apo structure, with a citrate molecule in the active site from the crystallization solution, overlays very well with the design model (data not shown), confirming that the mutations introduced during evolution did not affect the protein fold (Fig. 4a). Ser112 and Trp113 most likely increase the rigidity of a loop near the active site that is flexible in the OSH55.4 structure, Ser112 forms an intraloop backbone hydrogen bond, and Trp113 packs in the core of the protein. His126 and Tyr155 form a hydrogen bond that keeps helices 1 and 2 apart and may provide better access for the ligand (Supplementary Fig. 13a). In the bound structure of OSH55.4\_1 with FP<sub>pyne</sub> at 1.6-Å resolution, the entire FP<sub>pyne</sub> probe is visible in the structure, and the probe is covalently bound to Ser151, as expected (Fig. 4b). However, because of a slight rotation ( $\sim 20^\circ$  around the serine O $\gamma$ -phosphorus bond) of the organophosphate relative to the design model, water molecules interact with the organophosphate oxygen (O2 in Fig. 1a) rather than the designed oxyanion hole; this may be a shortcoming of the design or may be because of a reorientation that occurs after the reaction is completed (the structure is of the end product of the reaction rather than the transition state complex). Water molecules have been suggested to contribute to oxyanion holes in ketosteroid isomerases<sup>21</sup>. The structure also shows that the alkyl linker of the FP<sub>pyne</sub> probe packs tightly into a hydrophobic groove in the protein (Supplementary Fig. 13b), suggesting that improved probe binding likely contributes to the enhanced fluorophosphonate reactivity of OSH55.4\_1.



**Figure 4 | Crystal structures of OSH55.4\_1 bound to DFP.** (a) Crystal structure of the apo form of OSH55.4\_1 with a citrate molecule (white) from the crystallization buffer cocktail making extensive interactions with active site residues. (b) Crystal structure of OSH55.4\_1 covalently modified at Ser151 with FPyne (pink). (c) Crystal structure of OSH55.4\_1 covalently modified at Ser151 with DFP (red and blue). (d) Simulated annealing omit map around Ser151 ( $F_o - F_c$ ). Density is clearly evident on the serine, but the isopropyl groups of DFP are not distinguishable, probably owing to free rotation about the P-C bond.

To rule out the possibility that the loss of fluorophosphonate labeling of the H146A mutant was due to disrupted probe binding in the hydrophobic groove, we tested two additional mutants, H146I and H146V, that better preserved the hydrophobic properties of the groove. Both of the mutants notably decreased fluorophosphonate reactivity, consistent with a direct role for His146 in Ser151 activation (Supplementary Fig. 14).

A potential application of organophosphate-reactive proteins is the bioremediation and scavenging of toxic organophosphate-based pesticides and nerve agents<sup>22–24</sup>. To test whether our designs are able to react with organophosphates other than the activity-based fluorophosphonate probes, we solved the structure of OSH55.4\_1 in complex with diisopropyl fluorophosphate (DFP), a potent neurotoxin and serine protease inhibitor that mimics some of the structural features of more toxic chemical warfare agents<sup>25,26</sup>. Even though OSH55.4\_1 was neither designed nor evolved to specifically bind DFP, we found that DFP indeed forms a covalent adduct with the active site Ser151 (Fig. 4c,d) that is very similar to the complex formed by the fluorophosphonate moiety of the FPyne probe (PDB code 4JVV).

## DISCUSSION

The accuracy and the activity of the *de novo* designed catalytic triad active sites reported in this paper are major steps forward for computational enzyme design. The structures of the catalytic triad and quartet designs match the design models much more faithfully than the simpler hydrogen bond networks in previous computational designs<sup>12,27</sup>. The activation of serine, which has a  $pK_a$  of ~13, was markedly harder to achieve than the activation of cysteine ( $pK_a$  ~8) and histidine ( $pK_a$  ~7) in previous successful designs. Success in

activating the serine nucleophile most likely reflects the constraining of the catalytic triad through extensive packing interactions with the rest of the protein; the active site networks in these designs are considerably more buried than in most previous computationally designed enzymes. The control of histidine side chain conformation in the active designs described here contrasts with the lack of control found in the design of cysteine-histidine-based esterases<sup>10</sup>. In addition to burial of the histidine side chain, correct positioning is most likely facilitated by the greater strength of hydrogen bonding between histidine and serine compared to histidine and cysteine.

Together, the disrupted catalytic sites in two structures of inactive designs and the success with the catalytic side chain Boltzmann weight in distinguishing active from inactive designs (Supplementary Fig. 7) indicate the importance of catalytic triad preorganization for serine activation. The success with the OSH55 series of designs suggests that focusing design efforts on relatively buried sites built primarily on regular secondary structure elements is a good strategy for precisely controlling catalytic side chain conformation. Surface-exposed sites mounted on loop regions are likely to have many alternative conformations with roughly equal energies, making precise control over side chain conformation much harder to achieve.

The most active design in this study, OSH55.4\_1, uses a serine-histidine-histidine triad, which differs from the serine-histidine-aspartic acid/glutamic acid triads observed in most natural serine hydrolases. Parallels in nature include herpes virus peptidases with a serine-histidine-histidine triad<sup>17</sup>, but the contribution of the secondary histidine to catalysis appears to be smaller in these enzymes compared to OSH55.4\_1, where mutation of the secondary histidine reduces fluorophosphonate reactivity substantially. Also, unlike most serine esterases, the histidine in the OSH55 series of designs hydrogen bonds the catalytic serine through the N $\delta$  rather than the N $\epsilon$ . In contrast, the histidine in the structure of an inactive design (OSH97) interacts with the serine via the N $\epsilon$  (Supplementary Fig. 6c); evidently, the hydrogen bonding orientation of the histidine in the triad is not a major determinant for activation of the serine nucleophile.

Stoichiometric scavenging of organophosphates by injecting human butyrylcholinesterase (BCHE) into the bloodstream of poisoned individuals has shown promise in reducing toxicity from nerve agent exposure<sup>28</sup>. However, owing to the high molecular weight of BCHE, large amounts are required for stoichiometric inhibition. For example, 350 mg of human BCHE is required for every 1 mg of cyclosarin<sup>29</sup>. Another potential disadvantage of BCHE is unwanted hydrolytic cleavage of endogenous esters such as acetylcholine, leading to the imbalance of these metabolites in blood. The designed proteins described in this paper have potential advantages over BCHE for organophosphate scavenging in that they are much smaller proteins (17 kDa versus 65 kDa) and are likely to have limited hydrolytic activity against endogenous ester metabolites.

We have shown that computational methods can produce authentic catalytic triads that, with directed evolution, can achieve organophosphate reactivity at levels comparable to those of natural enzymes; starting with properly arrayed active site hydrogen bond networks, improvements in activities could be accomplished relatively quickly. A next challenge is to endow these and similarly designed catalytic triads with substrate recognition and other features required for proficient hydrolytic activity, particularly the capability to undergo repeated cycles of substrate acylation and deacylation. A separate and more practical challenge is to optimize the current designs to bind specific nerve agents such as Sarin, Tabun or VX for organophosphate scavenging and detection applications. More generally, the current work demonstrates the utility of breaking down the complexity of native enzyme catalysis into simpler subproblems that can be solved independently.

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## METHODS

Methods and any associated references are available in the [online version of the paper](#).

**Accession codes.** PDB: atomic coordinates and structure factors have been deposited under accession codes [4JVV](#), [4JLL](#), [4JCA](#), [4ESS](#), [4ETJ](#), [4ETK](#), [3V45](#), [3TP4](#), [4F2V](#) and [4DRT](#).

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## Author contributions

S.R., C.W., B.F.C. and D.B. conceived the project. S.R. and F.R. performed the computational design, S.R. expressed and purified the designed proteins. S.R. and K.Y. performed the yeast display experiments. C.W. performed the ABPP screening and mass spec experiments. C.W. and M.L.M. performed the fluoopol experiments. A.P.K., A.E.M., S.L., J.S., M.S. and J.F.H. performed the crystallography experiments. S.R., C.W., B.F.C. and D.B. analyzed data and wrote the manuscript.

## Competing financial interests

The authors declare no competing financial interests.

## Additional information

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## ONLINE METHODS

**Computational design.** Transition state models for *syn* and *anti* attack of the nucleophilic serine to the FP ligand were constructed using data from the previously published results<sup>14</sup>, and an ensemble of ligand conformers was then generated using the Omega software (OpenEye)<sup>31</sup>. The bond length, angle and dihedrals describing the theozyme geometry were adapted from previous findings<sup>30</sup>. The geometric parameters used to describe the theozyme are presented in **Supplementary Table 1**. Some of the constraints were loosened so that the number of initial matches (matched output) could be maximized. Geometric constraints for the oxyanion hole were taken from native serine hydrolases, as described in **Supplementary Table 1**. The final theozyme comprised a conformer library of the substrate in the transition state, the serine-histidine-aspartic acid/glutamic acid triad and one oxyanion-hole contribution from backbone amide protons. The theozyme shown in **Figure 1b** is more commonly observed in native proteases, but we did not restrict only to this theozyme arrangement during the matching process by RosettaMatch. In the geometric constraints used for Rosetta matching, either N $\epsilon$  or N $\delta$  of histidine is allowed to form a hydrogen bond with serine-O $\gamma$ , and the other unsatisfied N atom is then allowed to coordinate with either O $\epsilon$ 1 or O $\epsilon$ 2 of an aspartic acid or glutamic acid. This allows the alternate configuration of the theozyme present in the design to be realized as seen in OSH55. In the structures of an inactive design, OSH97, both the design and crystal structures have the theozyme geometry shown in **Figure 1b** (**Supplementary Fig. 6**).

Positions in protein scaffolds where the theozymes could be realized were identified using the RosettaMatch algorithm<sup>15</sup>. Catalytic interactions (imposed by theozyme geometry) were optimized by three rounds of sequence design and gradient-based energy minimization of the matches. During this minimization, restraints were added to the energy function to favor the desired theozyme geometry. Finally, the designed structures were repacked without the catalytic restraints. The designs to be tested experimentally were obtained by filtering for the following criteria: (i) the active site must be preorganized, where the r.m.s. deviation between idealized theozyme geometry and the active site in the final structures is less than 3 Å across all of the residues; (ii) the ligand binding energy is less than -4.0 Rosetta energy units (REU); and (iii) there are no more than two unsatisfied, buried polar ligand atoms. Out of ~380 designs that passed these filters, 100-top scoring designs were selected, of which 85 designs with unique scaffolds (**Supplementary Table 2**) were chosen for experimental characterization. For the second round of optimization, the crystal structure of OSH55 (PDB code 3V45) was used as a starting model. Scaffold residues within 15 Å of the active site residues were considered as design shell residues. Ser151 and His146 were not allowed to change during this round of design. Designs were filtered for additional interactions with His146, thereby stabilizing its designed rotamer. The Boltzmann rotamer probability distribution was calculated as previously described<sup>32</sup>. In our original design calculations, we did require that the desired active site conformation be the lowest energy state, but we did not impose a large energy gap between this (catalytically competent) and other conformations to bias it to be highly populated.

**Screening designs by gel-based ABPP.** FP-Rh, FP-biotin, FPpyne probes and azide-rhodamine were previously synthesized in-house with >99% purification, and aliquots from the more concentrated lab stocks were diluted to be used in the current study. Genes corresponding to the 85 designs were ordered from Genescript, USA. Genes in the pet29b vector were then expressed in *E. coli* BL21 cells in a 2-ml culture. One of the main advantages of ABPP is that the reactivity of each design can be assessed directly from the cell lysate in parallel without the laborious task of purifying a large number of proteins<sup>9</sup>. In a typical batch of screening, *E. coli* cells expressing the designed proteins were pelleted down and lysed in 0.5 ml of ice-cold PBS buffer by sonication. Soluble lysates were obtained by spinning at 14,000 r.p.m. in a desktop centrifuge for 30 min at 4 °C, and protein concentrations were adjusted to 0.5 mg/ml. 50  $\mu$ l of each lysate was labeled with 1  $\mu$ M of FP-Rh, FP-biotin or FPpyne probes for 1 h at room temperature (1  $\mu$ l of 50  $\mu$ M stock in DMSO). When the FPpyne probe was used, copper-catalyzed alkyne-azide cycloaddition ('click chemistry')<sup>33</sup> was performed to conjugate a fluorescent reporter tag to the probe-labeled proteins by the addition of 50  $\mu$ M of azide-rhodamine (1  $\mu$ l of a 2.5-mM stock in DMSO), 1 mM TCEP (1  $\mu$ l of a fresh 50-mM stock in water), 100  $\mu$ M ligand TBTA (3  $\mu$ l of a 1.7-mM stock in DMSO:*t*-butanol 1:4) and 1 mM CuSO<sub>4</sub> (1  $\mu$ l of a 50-mM stock in water). Samples were allowed to react at room

temperature for 1 h before 2 $\times$  gel loading buffer was added to quench the probe labeling/click-chemistry reaction. 12.5  $\mu$ g of each lysate was separated by an in-house 10% SDS-PAGE long gel, and in-gel fluorescence was visualized using a Hitachi FMBio II flatbed laser-induced fluorescence scanner (MiraiBio, Alameda, CA). For the FP-biotin labeled samples, immunoblotting was performed with IRDye Streptavidin (1:10,000) and scanned by an Odyssey imaging system (LI-COR). The gels were stained afterwards with Coomassie blue to assess the expression and relative abundance of each design. About 50% of the designs did not overexpress in soluble fraction, as shown by visual examination of the Coomassie blue-stained gel. For those designs that did show overexpression, their FP reactivity were ranked on the basis of fluorescence intensity normalized by the intensity from Coomassie blue staining, and designs exhibiting considerable probe labeling ('hits') were chosen for further characterization. An example of fluorescence and Coomassie blue-stained gel images is shown in **Supplementary Figure 1b**, and FP probe structures are shown in **Supplementary Figure 2**.

For the identified hits, knockout mutants of the active site residues were made using the Multi-site Lightning kit (Stratagene). After their sequences were confirmed, the mutants were subjected to ABPP screening along with their wild-type designs to confirm the specific probe labeling on the designed active site serine.

**Protein expression and purification.** Positive designs identified by ABPP were grown in 1 l of LB medium at 37 °C until induction (OD ~0.6), and the cells were switched to 22 °C to continue growing overnight. The harvested cell pellets were resuspended in the lysis buffer, which contained 1 $\times$  PBS buffer, 300 mM NaCl and 1 mM TCEP. No protease inhibitors were added in order to prevent preinhibition in the designed active sites. 1 mg/ml lysozyme was added right before the sonication step. After the lysate was centrifuged at 4,000g for 20 min, the soluble fraction was applied to 5 ml of TALON resin (Clontech) and then washed with 20 mM imidazole three times. The protein bound to the resin was eluted with 20 ml of the lysis buffer supplemented with 250 mM imidazole, and the eluted fraction was then further purified by gel filtration (HiLoad 26/60 Superdex 75) chromatography. The purified proteins were dialyzed overnight against the dialysis buffer containing 1 $\times$  PBS, 100 mM NaCl and 1 mM TCEP overnight, and its purity was confirmed by SDS-PAGE. Aliquots of the purified protein were flash frozen in liquid nitrogen and stored in -80 °C for further use.

**LC/MS/MS assay to identify serine-FP adducts.** Each of the purified protein designs (20  $\mu$ M, 50- $\mu$ l total volume) was incubated with DMSO or FPpyne (50  $\mu$ M) for 60 min at room temperature. The labeled protein samples were then prepared according to the in-solution trypsin digestion protocol published previously<sup>34</sup>. Briefly, the proteins were denatured in 6 M urea (150  $\mu$ l of 8 M urea in PBS was added), reduced by 10 mM of dithiothreitol for 30 min at room temperature (10  $\mu$ l of 200 mM stock in water was added) and alkylated by 20 mM iodoacetamide for 30 min at room temperature in the dark (10  $\mu$ l of 400 mM stock in water was added). The samples were diluted with ammonium bicarbonate (25 mM, 400  $\mu$ l) to 2 M urea and subjected to trypsin digestion (Promega; 4  $\mu$ l of 0.5  $\mu$ g/ $\mu$ l) overnight at 37 °C in the presence of 2 mM CaCl<sub>2</sub>. Digested peptide samples were desalted using the MacroSpin columns (The Nest Group, Inc.), concentrated and resuspended in 20  $\mu$ l of Buffer A (95% water, 5% acetonitrile, 0.1% formic acid). A 10- $\mu$ l aliquot was pressure-loaded onto a 100- $\mu$ m (inner diameter) fused silica capillary column (Agilent) with a 5- $\mu$ m tip that contained 10 cm C18 resin (Aqua 5  $\mu$ m, Phenomenex). LC/MS/MS analysis was performed on an LTQ-Orbitrap mass spectrometer (Thermo Scientific) coupled to an Agilent 1100 series HPLC. Peptides were eluted from the column using a 125-min gradient of 5–100% Buffer B (20% water, 80% acetonitrile, 0.1% formic acid). The flow rate through the column was 0.25  $\mu$ l/min, and the spray voltage was 2.5 kV. The LTQ was operated in data-dependent scanning mode, with one full MS scan (400–1,600 *m/z*) followed by MS/MS scans of the seven most abundant ions with dynamic exclusion enabled. The MS data was analyzed by SEQUEST<sup>35</sup> using a Uniprot *E. coli* sequence database (as of 20 July 2010) supplemented with the FASTA sequences of the designed proteins. A differential modification on serine of 371.22255 Da was defined in the search, and after the adducted peptides were identified, the corresponding MS1 chromatographic traces were extracted with  $\pm$ 15 p.p.m. accuracy from both the FPpyne- and DMSO-treated samples (**Fig. 3b** and **Supplementary Fig. 5**).

**Yeast display selection.** Libraries of OSH55.4 were constructed by randomizing six positions around the active site (112,113,126,130,155,159) using NNN oligos that were then assembled to make the full-length sequence. DNA libraries and linearized PETCON vectors were then transformed into EBY100 cells using the standard yeast display protocol<sup>18</sup>. The complexity of library was determined to be  $\sim 10^6$ . Cells were resuspended in SDCAA medium ( $10^7$  cells/ml) and were grown at 30 °C overnight. Cells were then centrifuged and induced at 22 °C in SGCAA medium for 24–48 h. Cells were labeled with 2  $\mu$ M of FP-biotin in PBSF buffer, washed with PBSF, secondary labeled with SAPE (Invitrogen) and anti-cMyc FITC (Miltenyi Biotech) and sorted by fluorescent gates (BD Influx sorter). Four rounds of sorting were carried out, after which the sequences converged to a unique sequence (**Supplementary Fig. 8**).

**Fuopool assays.** The fuopool assay was performed in a 384-well format and was based on a modified method published previously<sup>19</sup>. Briefly, purified OSH55.4\_1 (wild-type and the active site serine knockout mutant) as well as the native serine hydrolases RBBP9, PME1 and MAGL were diluted in assay buffer containing 50 mM Hepes, pH 7.5, 150 mM NaCl and 0.01% Pluronic F-127 (Invitrogen) to a series of concentrations ranging from 0.1  $\mu$ M to 5  $\mu$ M. 10  $\mu$ l of each enzyme or blank assay buffer was added to each well, and 1.1  $\mu$ l of FP-Rh probe (1.0  $\mu$ M stock in assay buffer and 0.1  $\mu$ M final concentration) was added to all wells by an automatic sample dispenser. The plates were read on an Envision plate reader (PerkinElmer) to measure fluorescence polarization signals at 3-min intervals for 45 min or longer. Each condition was represented by eight replicate wells, of which the averaged fuopool signal ( $\pm$ s.d.) was plotted in **Supplementary Figure 12a–e**. The time-dependent fuopool signals for each enzyme at each concentration were fitted into a one-phase association model  $Y = Y_0 + (plateau - Y_0) \times (1 - \exp(-k_{obs} \times X))$  in Prism 6 (GraphPad Software) to obtain  $Y_0$ , plateau and the observed rate constant  $k_{obs}$ . The fitted  $k_{obs}$  value was normalized by the enzyme concentration  $[E]$ , and the mean  $\pm$  s.e.m. of  $k_{obs}/[E]$  for each enzyme is reported in **Figure 3c**.

**Structure determination.** Protein crystallization and data collection were carried out in the Northeast Structural Genomics Consortium (NESG). Proteins were shipped to NESG, where they were assigned with NESG target identifiers (**Supplementary Table 4**). The proteins were subjected to another round of quality control that includes light scattering experiments to confirm the monodispersity of samples and MALDI-TOF analysis to confirm the molecular mass. pET expression vectors for these proteins have been deposited in the PSI Materials Repository (<http://psimr.asu.edu/>). The FP-alkyne complex of OSH55.4\_1 was made by incubating a twofold molar excess of FP-alkyne with OSH55.4\_1 for 1 h, and the unreacted FP-alkyne probe was removed by gel filtration. The protocol for the DFP complex will be made available upon request per biosafety requirement.

**Crystallization and optimization of crystals.** Initial crystallization conditions for the proteins were found by high-throughput robotic screening of 1,536 different conditions at the Hauptmann Woodward Institute, Buffalo, NY<sup>36</sup>. The hits were then used for laboratory optimization to grow crystals suitable for X-ray analysis. Crystals were obtained by the method of either microbatch under oil or vapor diffusion sitting drop. A 1- to 3- $\mu$ l protein solution drop

(8–10 mg/ml protein in 10 mM Tris, pH 7.4, 100 mM sodium chloride, 0.02% sodium azide and 5 mM dithiothreitol) was mixed with 1  $\mu$ l of the crystallization cocktail solution, and the mixture was incubated at 4 °C for several days to allow for crystal formation. The crystals were cryo-protected using 15–20% glycerol, ethylene glycol or *N*-paratone. Detailed crystallization conditions for each structure are available in **Supplementary Table 4**.

**X-ray data collection, processing and structure determination.** Single anomalous dispersion (SAD) X-ray data were collected at the wavelength 0.979 Å from single crystals under a temperature of 100 K on beamline X4A or X4C at the National Synchrotron Light Source at Brookhaven National Laboratory. The diffraction images were processed with HKL2000 and scaled with SCALEPACK<sup>37</sup>. Structures were solved by molecular replacement (MR) using BALBES<sup>38</sup> with the structure of a conserved hypothetical protein, TTHA0132 from *Thermus thermophilus* HB8 (PDB code 2DX6), as the search model. The MR models were refined with PHENIX<sup>39</sup> and then manually corrected with COOT<sup>40</sup> by analyzing  $2F_o - F_c$  and  $F_o - F_c$  electron density maps. Each of the current and final models was refined with translation, libration and screw-rotation (TLS) displacement of a pseudo-rigid body<sup>41</sup>. For the complex structures of OSH55.4\_1 with (FP-alkyne) and (DFP), the program eLBOW (<http://www.phenix-online.org/documentation/elbow.htm>) was used to generate crystallographic input files for model refinement and correction by COOT. Data collection and refinement statistics are reported in **Supplementary Table 5**.

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